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## Avinashilingam Institute for Home Science and Higher Education for Women

(Deemed to be University under Category 'A' by MHRD, Estd. u/s 3 of UGC Act 1956)

Re-accredited with 'A+' Grade by NAAC. Recognised by UGC Under Section 12B

Coimbatore - 641 043, Tamil Nadu, India

### Bachelor's Degree Examination – June 2021 VI Semester

Class : III UG  
Major : Physics

Time : 3 hours  
Max. Marks: 100

#### 18BPHC23 Quantum Mechanics and Relativity

##### Part A

10 x 1 = 10

##### Choose the Correct Answer

- If the positional inaccuracy of a particle is of the order of  $10^{-8}$  m, then the uncertainty in its momentum is \_\_\_\_\_  
a.  $10^{-26}$ m      b.  $10^{-16}$ m      c.  $10^{-6}$ m      d.  $10^{36}$ m      CO1 K 3
- The deBroglie wavelength of an electron whose speed is  $10^8$  m/s is  
a.  $7.28 \times 10^{-12}$ m      b.  $6.28 \times 10^{-12}$ m      c.  $7.28 \times 10^{-11}$ m      d.  $7.28 \times 10^{-10}$ m      CO1 K 3
- The quantum mechanical operator for energy is  
a.  $-i\hbar \frac{\partial}{\partial t}$       b.  $-i\hbar \nabla$       c.  $i\hbar \nabla$       d.  $i\hbar \frac{\partial}{\partial t}$       CO2 K1
- Different wave function with different quantum numbers may have the same energy value. Such a condition is called as  
a. Non-degeneracy states      b. Degeneracy state      c. Dependent states      d. none      CO2 K1
- The normalized wave function of the particle in a box of width L is  
a.  $\psi_n = \sqrt{\frac{3}{L}} \sin \frac{n\pi x}{L}$       b.  $\psi_n = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L}$   
c.  $\psi_n = \sqrt{\frac{L}{3}} \sin \frac{n\pi x}{L}$       d.  $\psi_n = \sqrt{\frac{L}{2}} \sin \frac{n\pi x}{L}$       CO3 K1
- Potential well of a linear harmonic oscillator is  
a. ellipse      b. circle      c. parabola      d. triangle      CO3 K1
- Galilean transformation shows that  
a. mass is invariant      b. acceleration is invariant  
c. time is invariant      d. all the above      CO4 K2
- The clock in a moving frame of reference will appear to go \_\_\_\_\_ than the clocks in the stationary frame of reference.  
a. Slower      b. faster      c. almost equal      d. none of the above      CO4 K2
- If the mass of the moving particle is tripled its rest mass, then its speed is \_\_\_\_\_  
a.  $0.9c$       b.  $0.94c$       c.  $c$       d.  $0.1c$       CO5, K3
- $1\text{\AA} =$  \_\_\_\_\_  
a.  $10^{-9}$ m      b.  $10^{-10}$ m      c.  $10^{-11}$ m      d.  $10^{-12}$ m      CO5, K1

**Part B**  
**Answer ALL questions**  
**Each answer should not exceed 400 words or two pages**

**5 x 6 = 30**

- 11.a. State Heisenberg uncertainty Principle and explain with any one illustration. CO1 K4  
(or)
- 11.b. Describe G.P Thomson's Experiment. CO1 K2
- 12.a. Explain the Physical significance of the wave function. CO2 K2  
(or)
- 12.b. Explain the postulates of Quantum Mechanics. CO2 K2
- 13.a. Derive Schrodinger wave equation for the Hydrogen atom . CO3 K3  
(or)
- 13.b. Find the probability that a particle trapped in a box L wide can be found between 0.45L and 0.55L for the ground and the first excited states. CO3 K3
- 14.a. State and deduce the mathematical expression for the law of addition of relative velocities. CO4 K4  
(or)
- 14.b. Describe the Lorentz – Fitzgerald length contraction. CO4K2
- 15.a. Explain minikowski's four dimensional space-time continuum. CO5 K2  
(or)
- 15.b. Explain the following terms, CO5 K2  
a. Gravitational red shift  
b. Black hole

**Part C**  
**Answer ALL questions**  
**Each answer should not exceed 800 words or four pages**

**5 x 12 = 60**

- 16.a. Describe Davisson and Germer experiment to confirm the existence of De Broglie waves. CO1 K2  
(or)
- 16.b. Define group velocity. Obtain an expression for phase and group velocities of de Broglie waves. CO1 K3
- 17.a. Derive time dependent Schrödinger's equation. What is the meaning of wave function? CO2 K4  
(or)
- 17.b. Derive time independent Schrödinger's equation. CO2 K4
- 18.a. A particle is bound in one dimensional infinite potential well. Determine the energy eigen values and energy eigen functions. Sketch the form of the wave function for  $n = 1, 2, 3$ . CO3 K4  
(or)
- 18.b. Explain the application of Schrodinger wave equation in linear harmonic oscillator. CO3 K4
- 19.a. Describe the Michelson-Morley experiment and explain the physical significance of negative results. CO4 K4  
(or)
- 19.b. State the postulates of special theory of relativity and Derive Lorentz transformation equations. CO4 K4
- 20.a. Derive Einstein's mass energy relation. CO5 K3  
(or)
- 20.b. Elaborate the general theory of relativity. CO5 K3

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