

**Avinasilingam Institute for Home Science and Higher Education
Coimbatore-641043**

**Master's Degree Examination -November-2017
III Semester**

**Class: II PG
Major: Chemistry**

**Max. Marks: 60
Time: 3 Hours**

12MCHC14-Polymer Chemistry

Part- A

(10 x 1/2 = 5)

- (1) Which of the following alkenes is most reactive towards cationic polymerization?
(a) $\text{CH}=\text{CHCH}_3$ (b) $\text{H}_2\text{C}=\text{CHCl}$ (c) $\text{H}_2\text{C}=\text{CHC}_6\text{H}_5$ (d) $\text{H}_2\text{C}=\text{CHCO}_2\text{CH}_3$
- (2) Which of the following act as inhibitor in vinyl polymerization?
(a) P-Benzoquinone (b) Benzophenone
(c) Phthalic anhydride (d) Benzoyl peroxide
- (3) A polymer which is used as a suture material, for stitching of wound after operation is
(a) PHBV (b) Nylon-2-Nylon-6 (c) Dextron (d) Dacron
- (4) Synthetic human hair wigs are made from a copolymer of vinyl chloride and acrylonitrile and called as
(a) PVC (b) Polyacrylonitrile (c) Cellulose (d) Dynel
- (5) The process involving heating of rubber with sulphur is known as
(a) Galvanization (b) Vulcanization (c) Besemerisation (d) Sulphonation
- (6) If N_1, N_2, N_3, \dots are the number of molecules with molecular masses M_1, M_2, M_3, \dots respectively then mass average mass is expressed as
(a) $\frac{\sum N_i M_i^2}{\sum N_i M_i}$ (b) $\frac{\sum N_i M_i}{\sum N_i}$ (c) $\frac{\sum M_i^2}{\sum N_i}$ (d) $\frac{\sum N_i M_i}{\sum M_i}$
- (7) The monomer units of silicones, a water repellent, acid resistant, and heat resistant polymer is
(a) Si (b) SiO_2 (c) R_2SiO (d) None of these
- (8) The Catalyst used for the polymerization of olefins is
(a) Ziegler-Natta Catalyst (b) Wilkinson's Catalyst (c) Pd-Catalyst (d) Zeolite
- (9) The technique of determination of molecular weight of a polymer by the depression of freezing point is called
(a) Osmometry (b) Cryoscopy (c) Viscometry (d) End group analysis
- (10). Which of the following polymers is used for insulation in electrical items and in the manufacture of gasket and valves?
(a) Bakelite (b) Polyaniline (c) Teflon (d) Polypyrrole

Part- B

(5 x 4 = 20)

Answer ALL questions

Each answer should not exceed 200 words (or) one page

- (11). (a) What are natural and synthetic polymers? Explain with suitable examples.
(OR)
(b) Write short notes on Ziegler Natta catalyst? Give its uses.
- (12). (a) Explain weight average molecular weight.
(OR)
(b) Write short notes on end group analysis with examples.
- (13). (a) Discuss the relationship between glass transition temperature and molecular weight.
(OR)
(b) What is thermal degradation of polymer? Explain with examples.
- (14). (a) Write short notes on (i) Plasticizers and (ii) Fillers.
(OR)
(b) Explain with neat sketch on fiber processing of polymers.
- (15). (a) Discuss about electroluminescent polymers.
(OR)
(b) Write a short note on biomedical polymers.

Part- C

(5 x 7 = 35)

Answer ALL questions

Each answer should not exceed 600 words (or) three pages

- (16). (a) What is degree of polymerization? Explain anionic and ring opening polymerization with examples.
(OR)
(b) Explain briefly on bulk and emulsion polymerization.
- (17). (a) Explain the following with suitable examples:
(i) Number average molecular weight and (ii) Viscosity average molecular weight
(OR)
(b) Explain in detail the colligative properties of polymers with suitable examples.
- (18). (a) What is glass transition temperature? Explain the relationship between T_g and T_m . Give the importance of T_g .
(OR)
(b) Discuss about high- energy degradation and oxidative degradation.
- (19). (a) Discuss the functions and examples of the following polymer additives:
(i) Heat stabilizers (ii) Flame retardants
(OR)
(c) Explain the compression moulding and blow moulding of plastics .
- (20). (a) Write short note on (i) Polyelectrolytes and (ii) Ionomers.
(OR)
(b) Give an account on liquid crystalline polymers and conducting polymers.
